

Electronic Supplementary Information for B506178D (Organic and Biomolecular Chemistry)

Amino acid conjugates of 1,1'-diaminoferrocene. Synthesis and chiral organisation

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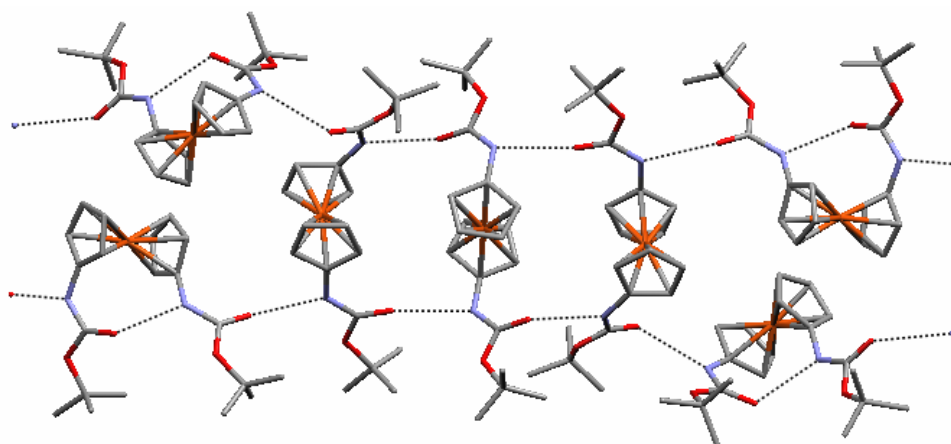


Fig. S1 Sheet like arrangement in the X-ray single crystal structure of compound **6** involving three different rotamers present in the asymmetric unit cell and their linkages through hydrogen bonding. Selected hydrogen bond distances O51...N41 2.709, N51A...O31 2.763, N51...O31A 2.763, O51A...N41A 2.709, O41...N11 2.781, N31...O21B 2.735, O11...N21 2.614, O21...N31B 2.735. [A =1-x,1-y,-z B=1.5-x,1/2+y,1/2-z].

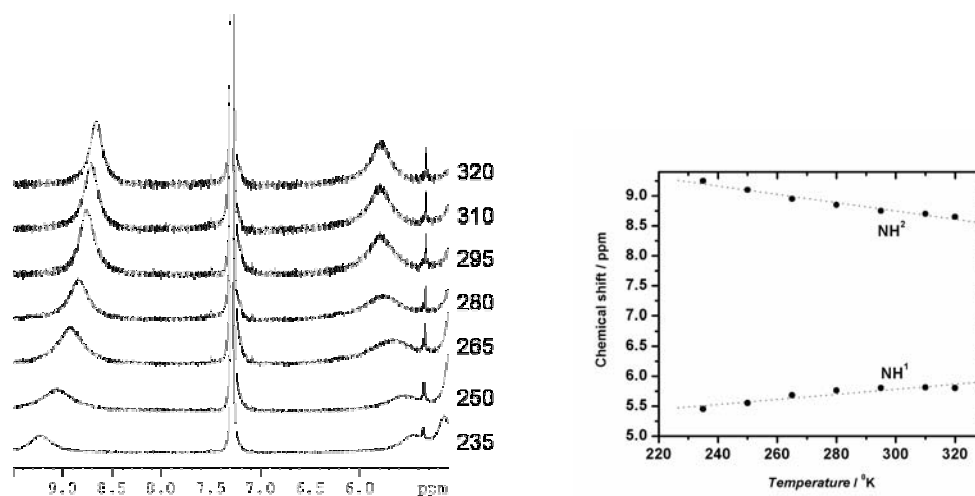


Fig. s2 Variable NMR study of 1,1'-bis(*tert*-butoxycarbonyl-L-alanine-amido) ferrocene (7), NH¹ is close to ferrocene and NH² is the one which is attached to Boc.

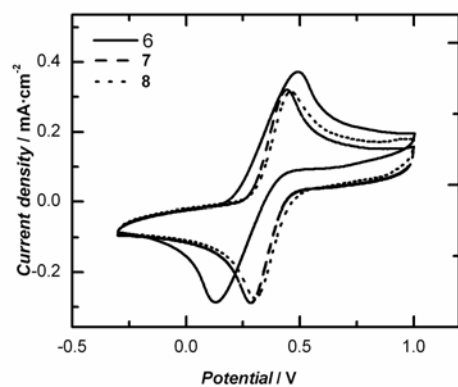


Fig. s3 Cyclic voltammogram SWV spectra of 0.1 M compounds **6**, **7**, **8** measured using GCE electrode vs. Ag/AgCl, scan rate 0.1 v/s, in CH₂Cl₂/0.1 M TBAP. The $E_{1/2}$ of the Fc/Fc⁺ couple under the experimental conditions is 448(+/-5) mV (vs. Ag/AgCl)

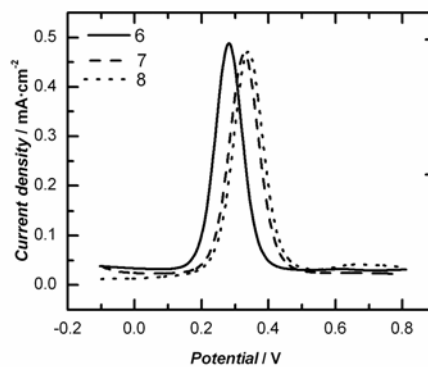


Fig. s4 SWV spectra of 0.1 M compounds **6**, **7**, **8** measured using GCE electrode vs. Ag/AgCl, scan rate 0.1 v/s, in CH₂Cl₂/0.1 M TBAP. The $E_{1/2}$ of the Fc/Fc⁺ couple under the experimental conditions is 448(+/-5) mV (vs. Ag/AgCl).

Structural data for compounds **6**, **7** and **8**.

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# SUBMISSION DETAILS

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  in support of an upcoming communication to be published in
  Organic and Biomolecular Chemistry.
  Please forward any questions concerning
  this CIF to Dr. Gabriele Schatte
  (gabriele.schatte@usask.ca).
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_publ_section_acknowledgements
; NSERC, CFI
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F2 > 2sigma(F2) is used only for calculating R-factors(gt) etc.
and is
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and R-
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C10 C 0.5247(2) -0.1194(3) 0.18278(7) 0.0241(7) Uani 1 1 d . . .
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H52 H 0.6427 0.3923 -0.0438 0.030 Uiso 1 1 calc . . .
C53 C 0.6004(2) 0.3468(3) 0.00500(7) 0.0258(7) Uani 1 1 d . . .
H53 H 0.6795 0.3439 0.0155 0.031 Uiso 1 1 calc . . .
C54 C 0.4936(2) 0.3258(3) 0.01997(7) 0.0242(7) Uani 1 1 d . . .
H54 H 0.4801 0.3064 0.0422 0.029 Uiso 1 1 calc . . .
C55 C 0.2299(2) 0.3428(3) 0.02462(7) 0.0236(7) Uani 1 1 d . . .
C56 C 0.0330(3) 0.3318(3) 0.04326(8) 0.0347(8) Uani 1 1 d . . .
C57 C 0.0200(3) 0.4669(3) 0.05105(9) 0.0470(10) Uani 1 1 d . . .
H57A H 0.0996 0.4964 0.0591 0.056 Uiso 1 1 calc R . .
H57B H -0.0428 0.4783 0.0677 0.056 Uiso 1 1 calc R . .
H57C H -0.0048 0.5071 0.0310 0.056 Uiso 1 1 calc R . .
C58 C 0.0617(3) 0.2572(3) 0.07352(9) 0.0491(10) Uani 1 1 d . . .
H58A H 0.0828 0.1805 0.0664 0.059 Uiso 1 1 calc R . .
H58B H -0.0112 0.2542 0.0874 0.059 Uiso 1 1 calc R . .
H58C H 0.1315 0.2897 0.0859 0.059 Uiso 1 1 calc R . .
C59 C -0.0714(3) 0.2797(4) 0.02605(10) 0.0658(13) Uani 1 1 d . . .
H59A H -0.0883 0.3208 0.0056 0.079 Uiso 1 1 calc R . .
H59B H -0.1446 0.2827 0.0398 0.079 Uiso 1 1 calc R . .

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H59C H -0.0518 0.2006 0.0211 0.079 Uiso 1 1 calc R . .

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O32 0.0264(11) 0.0335(13) 0.0283(12) 0.0020(10) -0.0018(9) -0.0076(9)
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O51 0.0162(10) 0.0474(14) 0.0176(11) 0.0006(10) -0.0006(8) 0.0009(9)
O52 0.0124(10) 0.0662(17) 0.0258(12) -0.0077(11) 0.0070(9) -0.0046(10)
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C19 0.0262(18) 0.069(3) 0.067(3) -0.012(2) 0.0113(18) 0.0024(18)
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All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

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Fe1 C11 2.051(3) . ?
Fe1 C20 2.063(3) . ?

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C34 Fe2 C42 104.65 (13) . . ?
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C40 Fe2 C33 116.53 (13) . . ?
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C53 Fe3 C51 117.37 (11) 3_665 . ?
C53 Fe3 C51 62.63 (11) . . ?
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C53 Fe3 C51 117.37 (11) . 3_665 ?
C51 Fe3 C51 180.0 . 3_665 ?
C52 Fe3 C50 119.09 (10) . 3_665 ?
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C53 Fe3 C50 120.71 (11) 3_665 . ?
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C50 Fe3 C50 180.0 3_665 . ?
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C35 O32 C36 124.8(2) . . ?
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C55 O52 C56 116.4(2) . . ?
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C10 N11 H11 118.4 . . ?
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C20 N21 H21 118.8 . . ?
C25 N21 H21 118.8 . . ?
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C35 N31 H31 115.4 . . ?
C30 N31 H31 115.4 . . ?
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N11 C10 C14 133.7(2) . . ?
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N11 C10 Fe1 123.4(2) . . ?
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C14 C13 Fe1 65.06(18) . . ?
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N11 C15 O12 107.1(3) . . ?
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O12 C16 C18 116.4(3) . . ?
C19 C16 C18 112.8(3) . . ?
O12 C16 C17 102.5(3) . . ?
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C18 C16 C17 105.1(3) . . ?
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H17A C17 H17C 109.5 . . ?
H17B C17 H17C 109.5 . . ?
C16 C18 H18A 109.5 . . ?
C16 C18 H18B 109.5 . . ?
H18A C18 H18B 109.5 . . ?
C16 C18 H18C 109.5 . . ?
H18A C18 H18C 109.5 . . ?
H18B C18 H18C 109.5 . . ?
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H19B C19 H19C 109.5 . . ?
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O22 C26 C27 108.5(3) . . ?
C29 C26 C28 108.9(3) . . ?
O22 C26 C28 114.6(2) . . ?
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C26 C28 H28B 109.5 . . ?
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C26 C28 H28C 109.5 . . ?
H28A C28 H28C 109.5 . . ?
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C26 C29 H29B 109.5 . . ?
H29A C29 H29B 109.5 . . ?
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C31 C32 Fe2 65.31(14) . . ?
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Fe2 C32 H32 131.7 . . ?
C32 C33 C34 104.1(3) . . ?

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C32 C33 Fe2 70.34(17) . . ?
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Fe2 C33 H33 130.9 . . ?
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C33 C34 Fe2 73.38(17) . . ?
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Fe2 C34 H34 124.9 . . ?
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O31 C35 N31 123.0(3) . . ?
O32 C35 N31 113.4(2) . . ?
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C41 C42 Fe2 67.09(16) . . ?
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Fe2 C42 H42 129.6 . . ?
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C49 C46 O42 101.9(3) . . ?
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H47A C47 H47C 109.5 . . ?
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H48B C48 H48C 109.5 . . ?
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H49B C49 H49C 109.5 . . ?
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C53 C52 H52 126.0 . . ?
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C54 C53 H53 124.3 . . ?
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C59 C56 C57 116.0(3) . . ?
C58 C56 C57 114.7(3) . . ?
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H57A C57 H57C 109.5 . . ?
H57B C57 H57C 109.5 . . ?
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C56 C58 H58C 109.5 . . ?
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C56 C59 H59B 109.5 . . ?
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C12 Fe1 C10 C11 -41.04 (19) ?
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C24 Fe1 C11 C10 -90.3 (2) ?
C22 Fe1 C11 C12 74.3 (3) ?
C10 Fe1 C11 C12 -113.7 (3) ?
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C20 Fe1 C11 C12 -169.6 (2) ?
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C13 Fe1 C20 N21 -52.3(3) ?
C24 Fe1 C20 N21 120.9(4) ?

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C22 Fe1 C20 C21 -42.3(2) ?
C10 Fe1 C20 C21 147.12(19) ?
C14 Fe1 C20 C21 99.4(2) ?
C11 Fe1 C20 C21 -176.1(2) ?
C23 Fe1 C20 C21 -89.9(2) ?
C12 Fe1 C20 C21 38.9(6) ?
C13 Fe1 C20 C21 62.1(2) ?
C24 Fe1 C20 C21 -124.7(3) ?
C22 Fe1 C20 C24 82.4(2) ?
C10 Fe1 C20 C24 -88.2(2) ?
C21 Fe1 C20 C24 124.7(3) ?
C14 Fe1 C20 C24 -135.91(18) ?
C11 Fe1 C20 C24 -51.4(3) ?
C23 Fe1 C20 C24 34.77(19) ?
C12 Fe1 C20 C24 163.6(5) ?
C13 Fe1 C20 C24 -173.18(17) ?
N21 C20 C21 C22 175.8(3) ?
C24 C20 C21 C22 -0.6(4) ?
Fe1 C20 C21 C22 60.7(2) ?
N21 C20 C21 Fe1 115.2(3) ?
C24 C20 C21 Fe1 -61.3(2) ?
C10 Fe1 C21 C22 -173.5(2) ?
C14 Fe1 C21 C22 151.4(2) ?
C20 Fe1 C21 C22 -110.8(3) ?
C23 Fe1 C21 C22 -38.3(2) ?
C12 Fe1 C21 C22 78.9(3) ?
C13 Fe1 C21 C22 113.0(2) ?
C24 Fe1 C21 C22 -76.4(2) ?
C22 Fe1 C21 C20 110.8(3) ?
C10 Fe1 C21 C20 -62.7(3) ?
C14 Fe1 C21 C20 -97.8(2) ?
C23 Fe1 C21 C20 72.54(19) ?
C12 Fe1 C21 C20 -170.31(18) ?
C13 Fe1 C21 C20 -136.18(18) ?
C24 Fe1 C21 C20 34.39(18) ?
C20 C21 C22 C23 0.3(4) ?
Fe1 C21 C22 C23 64.8(2) ?
C20 C21 C22 Fe1 -64.5(2) ?
C14 Fe1 C22 C21 -50.3(3) ?
C11 Fe1 C22 C21 -168.4(2) ?
C20 Fe1 C22 C21 41.8(2) ?
C23 Fe1 C22 C21 120.5(3) ?
C12 Fe1 C22 C21 -125.5(2) ?
C13 Fe1 C22 C21 -83.3(2) ?
C24 Fe1 C22 C21 86.1(2) ?
C21 Fe1 C22 C23 -120.5(3) ?
C14 Fe1 C22 C23 -170.74(19) ?
C11 Fe1 C22 C23 71.1(2) ?
C20 Fe1 C22 C23 -78.69(19) ?
C12 Fe1 C22 C23 114.02(19) ?
C13 Fe1 C22 C23 156.25(17) ?
C24 Fe1 C22 C23 -34.35(17) ?
C21 C22 C23 C24 0.2(4) ?
Fe1 C22 C23 C24 64.0(2) ?
C21 C22 C23 Fe1 -63.8(3) ?
C22 Fe1 C23 C24 -118.2(3) ?
C10 Fe1 C23 C24 68.9(3) ?

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C21 Fe1 C23 C24 -81.9(2) ?
C11 Fe1 C23 C24 108.7(2) ?
C20 Fe1 C23 C24 -38.9(2) ?
C12 Fe1 C23 C24 152.1(2) ?
C13 Fe1 C23 C24 -169.2(2) ?
C10 Fe1 C23 C22 -172.86(16) ?
C21 Fe1 C23 C22 36.38(17) ?
C11 Fe1 C23 C22 -133.11(17) ?
C20 Fe1 C23 C22 79.35(19) ?
C12 Fe1 C23 C22 -89.7(2) ?
C13 Fe1 C23 C22 -50.9(3) ?
C24 Fe1 C23 C22 118.2(3) ?
C22 C23 C24 C20 -0.5(4) ?
Fe1 C23 C24 C20 54.6(2) ?
C22 C23 C24 Fe1 -55.1(2) ?
N21 C20 C24 C23 -175.6(3) ?
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Fe1 C20 C24 C23 -56.7(2) ?
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C22 Fe1 C24 C23 40.8(2) ?
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C14 Fe1 C24 C23 -173.8(2) ?
C11 Fe1 C24 C23 -87.5(2) ?
C20 Fe1 C24 C23 118.8(3) ?
C12 Fe1 C24 C23 -54.4(4) ?
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C21 Fe1 C24 C20 -34.29(17) ?
C14 Fe1 C24 C20 67.4(2) ?
C11 Fe1 C24 C20 153.70(17) ?
C23 Fe1 C24 C20 -118.8(3) ?
C12 Fe1 C24 C20 -173.1(2) ?
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C26 O22 C25 N21 -165.0(3) ?
C20 N21 C25 O21 -12.1(6) ?
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C25 O22 C26 C29 167.9(3) ?
C25 O22 C26 C27 -78.0(3) ?
C25 O22 C26 C28 53.2(4) ?
C35 N31 C30 C31 -3.3(4) ?
C35 N31 C30 C34 175.4(3) ?
C35 N31 C30 Fe2 -99.0(3) ?
C34 Fe2 C30 C31 116.9(2) ?
C40 Fe2 C30 C31 -48.1(4) ?
C44 Fe2 C30 C31 -81.9(2) ?
C41 Fe2 C30 C31 163.4(3) ?
C42 Fe2 C30 C31 -162.16(19) ?
C33 Fe2 C30 C31 79.52(19) ?
C32 Fe2 C30 C31 37.99(18) ?
C43 Fe2 C30 C31 -121.99(19) ?
C40 Fe2 C30 C34 -165.0(3) ?
C44 Fe2 C30 C34 161.23(18) ?
C41 Fe2 C30 C34 46.6(4) ?
C31 Fe2 C30 C34 -116.9(2) ?
C42 Fe2 C30 C34 80.9(2) ?

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C33 Fe2 C30 C34 -37.38 (18) ?
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C43 Fe2 C30 C34 121.11 (19) ?
C34 Fe2 C30 N31 -118.2 (3) ?
C40 Fe2 C30 N31 76.9 (4) ?
C44 Fe2 C30 N31 43.1 (3) ?
C41 Fe2 C30 N31 -71.6 (5) ?
C31 Fe2 C30 N31 124.9 (3) ?
C42 Fe2 C30 N31 -37.2 (3) ?
C33 Fe2 C30 N31 -155.6 (3) ?
C32 Fe2 C30 N31 162.9 (3) ?
C43 Fe2 C30 N31 2.9 (3) ?
C34 C30 C31 C32 -1.2 (3) ?
N31 C30 C31 C32 177.8 (2) ?
Fe2 C30 C31 C32 -57.51 (17) ?
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N31 C30 C31 Fe2 -124.7 (3) ?
C34 Fe2 C31 C30 -41.86 (18) ?
C40 Fe2 C31 C30 160.05 (18) ?
C44 Fe2 C31 C30 112.65 (18) ?
C41 Fe2 C31 C30 -164.5 (3) ?
C42 Fe2 C31 C30 43.7 (4) ?
C33 Fe2 C31 C30 -88.62 (19) ?
C32 Fe2 C31 C30 -122.5 (2) ?
C43 Fe2 C31 C30 72.2 (2) ?
C34 Fe2 C31 C32 80.61 (18) ?
C40 Fe2 C31 C32 -77.5 (2) ?
C30 Fe2 C31 C32 122.5 (2) ?
C44 Fe2 C31 C32 -124.88 (18) ?
C41 Fe2 C31 C32 -42.1 (4) ?
C42 Fe2 C31 C32 166.2 (3) ?
C33 Fe2 C31 C32 33.85 (17) ?
C43 Fe2 C31 C32 -165.34 (17) ?
C30 C31 C32 C33 0.9 (3) ?
Fe2 C31 C32 C33 -55.11 (19) ?
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C41 Fe2 C32 C33 -72.5 (2) ?
C31 Fe2 C32 C33 124.3 (2) ?
C42 Fe2 C32 C33 -42.2 (4) ?
C43 Fe2 C32 C33 164.0 (3) ?
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C41 Fe2 C32 C31 163.18 (17) ?
C42 Fe2 C32 C31 -166.5 (3) ?
C33 Fe2 C32 C31 -124.3 (2) ?
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Fe2 C32 C33 C34 -52.95 (18) ?
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C40 Fe2 C33 C32 78.70 (19) ?

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C30 Fe2 C33 C32 -80.75 (18) ?
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C43 Fe2 C33 C32 -164.7 (3) ?
C40 Fe2 C33 C34 -162.47 (17) ?
C30 Fe2 C33 C34 38.09 (17) ?
C44 Fe2 C33 C34 163.8 (3) ?
C41 Fe2 C33 C34 -120.62 (18) ?
C31 Fe2 C33 C34 81.61 (18) ?
C42 Fe2 C33 C34 -79.1 (2) ?
C32 Fe2 C33 C34 118.8 (2) ?
C43 Fe2 C33 C34 -45.8 (4) ?
C31 C30 C34 C33 1.1 (3) ?
N31 C30 C34 C33 -177.9 (2) ?
Fe2 C30 C34 C33 61.0 (2) ?
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C32 C33 C34 C30 -0.5 (3) ?
Fe2 C33 C34 C30 -58.81 (19) ?
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C44 Fe2 C34 C30 -43.7 (4) ?
C41 Fe2 C34 C30 -162.72 (17) ?
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C32 Fe2 C34 C30 85.89 (18) ?
C43 Fe2 C34 C30 -74.7 (2) ?
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C30 Fe2 C34 C33 -122.6 (2) ?
C44 Fe2 C34 C33 -166.4 (3) ?
C41 Fe2 C34 C33 74.7 (2) ?
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C36 O32 C35 O31 -4.2 (4) ?
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C30 N31 C35 O32 -179.8 (2) ?
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C35 O32 C36 C39 -57.5 (3) ?
C35 O32 C36 C37 179.3 (2) ?
C45 N41 C40 C41 -33.1 (4) ?
C45 N41 C40 C44 150.8 (2) ?
C45 N41 C40 Fe2 61.2 (4) ?
C34 Fe2 C40 C41 43.7 (4) ?
C30 Fe2 C40 C41 -163.9 (3) ?
C44 Fe2 C40 C41 -120.1 (2) ?
C31 Fe2 C40 C41 160.84 (19) ?
C42 Fe2 C40 C41 -40.5 (2) ?
C33 Fe2 C40 C41 75.6 (2) ?
C32 Fe2 C40 C41 116.2 (2) ?
C43 Fe2 C40 C41 -84.7 (2) ?
C34 Fe2 C40 N41 -77.6 (4) ?

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C30 Fe2 C40 N41 74.7(4) ?
C44 Fe2 C40 N41 118.5(3) ?
C41 Fe2 C40 N41 -121.3(4) ?
C31 Fe2 C40 N41 39.5(3) ?
C42 Fe2 C40 N41 -161.8(3) ?
C33 Fe2 C40 N41 -45.7(3) ?
C32 Fe2 C40 N41 -5.2(3) ?
C43 Fe2 C40 N41 154.0(3) ?
C34 Fe2 C40 C44 163.8(3) ?
C30 Fe2 C40 C44 -43.8(4) ?
C41 Fe2 C40 C44 120.1(2) ?
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C42 Fe2 C40 C44 79.67(18) ?
C33 Fe2 C40 C44 -164.25(16) ?
C32 Fe2 C40 C44 -123.70(17) ?
C43 Fe2 C40 C44 35.45(16) ?
N41 C40 C41 C42 -176.3(2) ?
C44 C40 C41 C42 0.4(3) ?
Fe2 C40 C41 C42 59.05(18) ?
N41 C40 C41 Fe2 124.7(3) ?
C44 C40 C41 Fe2 -58.67(17) ?
C34 Fe2 C41 C40 -161.69(19) ?
C30 Fe2 C41 C40 162.2(3) ?
C44 Fe2 C41 C40 40.80(18) ?
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C33 Fe2 C41 C40 -118.3(2) ?
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C43 Fe2 C41 C40 83.0(2) ?
C34 Fe2 C41 C42 81.2(2) ?
C40 Fe2 C41 C42 -117.1(3) ?
C30 Fe2 C41 C42 45.1(4) ?
C44 Fe2 C41 C42 -76.3(2) ?
C31 Fe2 C41 C42 -164.7(3) ?
C33 Fe2 C41 C42 124.6(2) ?
C32 Fe2 C41 C42 163.35(19) ?
C43 Fe2 C41 C42 -34.12(19) ?
C40 C41 C42 C43 -0.1(3) ?
Fe2 C41 C42 C43 56.6(2) ?
C40 C41 C42 Fe2 -56.76(17) ?
C34 Fe2 C42 C43 117.42(18) ?
C40 Fe2 C42 C43 -86.83(19) ?
C30 Fe2 C42 C43 70.7(2) ?
C44 Fe2 C42 C43 -36.94(17) ?
C41 Fe2 C42 C43 -126.3(3) ?
C31 Fe2 C42 C43 38.4(4) ?
C33 Fe2 C42 C43 162.80(17) ?
C32 Fe2 C42 C43 -166.8(3) ?
C34 Fe2 C42 C41 -116.2(2) ?
C40 Fe2 C42 C41 39.51(18) ?
C30 Fe2 C42 C41 -162.92(18) ?
C44 Fe2 C42 C41 89.4(2) ?
C31 Fe2 C42 C41 164.8(3) ?
C33 Fe2 C42 C41 -70.9(2) ?
C32 Fe2 C42 C41 -40.5(4) ?
C43 Fe2 C42 C41 126.3(3) ?
C41 C42 C43 C44 -0.2(3) ?

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Fe2 C42 C43 C44 53.80 (18) ?
C41 C42 C43 Fe2 -54.0 (2) ?
C34 Fe2 C43 C44 160.51 (18) ?
C40 Fe2 C43 C44 -40.50 (18) ?
C30 Fe2 C43 C44 116.20 (19) ?
C41 Fe2 C43 C44 -84.82 (19) ?
C31 Fe2 C43 C44 75.8 (2) ?
C42 Fe2 C43 C44 -120.0 (2) ?
C33 Fe2 C43 C44 -164.3 (3) ?
C32 Fe2 C43 C44 45.1 (4) ?
C34 Fe2 C43 C42 -79.5 (2) ?
C40 Fe2 C43 C42 79.52 (19) ?
C30 Fe2 C43 C42 -123.78 (18) ?
C44 Fe2 C43 C42 120.0 (2) ?
C41 Fe2 C43 C42 35.20 (18) ?
C31 Fe2 C43 C42 -164.21 (17) ?
C33 Fe2 C43 C42 -44.3 (4) ?
C32 Fe2 C43 C42 165.1 (3) ?
C42 C43 C44 C40 0.4 (3) ?
Fe2 C43 C44 C40 54.81 (17) ?
C42 C43 C44 Fe2 -54.39 (18) ?
C41 C40 C44 C43 -0.5 (3) ?
N41 C40 C44 C43 176.1 (2) ?
Fe2 C40 C44 C43 -60.93 (19) ?
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C31 Fe2 C44 C43 -119.31 (19) ?
C42 Fe2 C44 C43 38.28 (18) ?
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C32 Fe2 C44 C43 -162.85 (18) ?
C34 Fe2 C44 C40 -164.7 (3) ?
C30 Fe2 C44 C40 162.43 (15) ?
C41 Fe2 C44 C40 -37.30 (16) ?
C31 Fe2 C44 C40 119.52 (16) ?
C42 Fe2 C44 C40 -82.89 (17) ?
C33 Fe2 C44 C40 42.9 (4) ?
C32 Fe2 C44 C40 75.99 (18) ?
C43 Fe2 C44 C40 -121.2 (2) ?
C46 O42 C45 O41 -7.5 (4) ?
C46 O42 C45 N41 171.8 (2) ?
C40 N41 C45 O41 6.1 (4) ?
C40 N41 C45 O42 -173.2 (2) ?
C45 O42 C46 C48 -63.1 (4) ?
C45 O42 C46 C49 -175.6 (3) ?
C45 O42 C46 C47 61.7 (3) ?
C55 N51 C50 C54 34.5 (5) ?
C55 N51 C50 C51 -149.2 (3) ?
C55 N51 C50 Fe3 -61.5 (4) ?
C52 Fe3 C50 N51 -153.0 (3) ?
C52 Fe3 C50 N51 27.0 (3) $\bar{3}$ _665 ?
C53 Fe3 C50 N51 -19.0 (3) $\bar{3}$ _665 ?
C53 Fe3 C50 N51 161.0 (3) ?
C51 Fe3 C50 N51 -115.3 (3) ?

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C51 Fe3 C50 N51 64.7(3) 3_665 . . . ?
C54 Fe3 C50 N51 -57.8(3) 3_665 . . . ?
C54 Fe3 C50 N51 122.2(3) . . . ?
C52 Fe3 C50 C54 84.78(18) . . . ?
C52 Fe3 C50 C54 -95.22(18) 3_665 . . . ?
C53 Fe3 C50 C54 -141.14(16) 3_665 . . . ?
C53 Fe3 C50 C54 38.85(16) . . . ?
C51 Fe3 C50 C54 122.5(2) . . . ?
C51 Fe3 C50 C54 -57.5(2) 3_665 . . . ?
C54 Fe3 C50 C54 180.0 3_665 . . . ?
C52 Fe3 C50 C51 -37.72(17) . . . ?
C52 Fe3 C50 C51 142.29(17) 3_665 . . . ?
C53 Fe3 C50 C51 96.36(18) 3_665 . . . ?
C53 Fe3 C50 C51 -83.65(18) . . . ?
C51 Fe3 C50 C51 180.0 3_665 . . . ?
C54 Fe3 C50 C51 57.5(2) 3_665 . . . ?
C54 Fe3 C50 C51 -122.5(2) . . . ?
N51 C50 C51 C52 -176.2(3) . . . ?
C54 C50 C51 C52 0.5(4) . . . ?
Fe3 C50 C51 C52 62.8(2) . . . ?
N51 C50 C51 Fe3 121.0(3) . . . ?
C54 C50 C51 Fe3 -62.2(2) . . . ?
C52 Fe3 C51 C52 -180.0 3_665 . . . ?
C53 Fe3 C51 C52 139.92(18) 3_665 . . . ?
C53 Fe3 C51 C52 -40.09(18) . . . ?
C50 Fe3 C51 C52 65.7(3) 3_665 . . . ?
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C54 Fe3 C51 C52 -80.38(19) . . . ?
C52 Fe3 C51 C50 114.3(3) . . . ?
C52 Fe3 C51 C50 -65.7(3) 3_665 . . . ?
C53 Fe3 C51 C50 -105.81(18) 3_665 . . . ?
C53 Fe3 C51 C50 74.19(18) . . . ?
C50 Fe3 C51 C50 180.0 3_665 . . . ?
C54 Fe3 C51 C50 -146.10(16) 3_665 . . . ?
C54 Fe3 C51 C50 33.90(16) . . . ?
C50 C51 C52 C53 -0.1(3) . . . ?
Fe3 C51 C52 C53 63.4(2) . . . ?
C50 C51 C52 Fe3 -63.5(2) . . . ?
C53 Fe3 C52 C51 -63.9(2) 3_665 . . . ?
C53 Fe3 C52 C51 116.1(2) . . . ?
C51 Fe3 C52 C51 180.0 3_665 . . . ?
C50 Fe3 C52 C51 -139.72(18) 3_665 . . . ?
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C54 Fe3 C52 C51 -98.35(19) 3_665 . . . ?
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C53 Fe3 C52 C53 180.0 3_665 . . . ?
C51 Fe3 C52 C53 -116.1(2) . . . ?
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C50 Fe3 C52 C53 104.14(17) 3_665 . . . ?
C50 Fe3 C52 C53 -75.86(17) . . . ?
C54 Fe3 C52 C53 145.51(16) 3_665 . . . ?
C54 Fe3 C52 C53 -34.49(16) . . . ?
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Fe3 C52 C53 C54 64.7(2) . . . ?
C51 C52 C53 Fe3 -65.0(2) . . . ?
C52 Fe3 C53 C54 -119.9(3) . . . ?

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C52 Fe3 C53 C54 60.1(3) 3_665 . . . ?
 C51 Fe3 C53 C54 -83.52(19) ?
 C51 Fe3 C53 C54 96.47(19) 3_665 . . . ?
 C50 Fe3 C53 C54 140.40(17) 3_665 . . . ?
 C50 Fe3 C53 C54 -39.60(17) ?
 C54 Fe3 C53 C54 180.0 3_665 . . . ?
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 C51 Fe3 C53 C52 -143.64(15) 3_665 . . . ?
 C50 Fe3 C53 C52 -99.72(17) 3_665 . . . ?
 C50 Fe3 C53 C52 80.29(17) ?
 C54 Fe3 C53 C52 -60.1(3) 3_665 . . . ?
 C54 Fe3 C53 C52 119.9(3) ?
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 Fe3 C53 C54 C50 61.2(2) ?
 C52 C53 C54 Fe3 -60.6(2) ?
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 C53 Fe3 C54 C53 -180.0 3_665 . . . ?
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 C51 Fe3 C54 C53 -102.09(19) 3_665 . . . ?
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 C53 Fe3 C54 C50 -113.3(2) ?
 C51 Fe3 C54 C50 -35.41(15) ?
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 C50 Fe3 C54 C50 180.0 3_665 . . . ?
 C56 O52 C55 O51 0.5(5) ?
 C56 O52 C55 N51 -179.0(3) ?
 C50 N51 C55 O51 0.6(5) ?
 C50 N51 C55 O52 -179.9(3) ?
 C55 O52 C56 C59 -174.5(3) ?
 C55 O52 C56 C58 -63.4(4) ?
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 _geom_hbond_site_symmetry_A
 N11 H11 O41 0.88 2.05 2.887(3) 158.8 .
 N21 H21 O11 0.88 1.99 2.753(3) 144.1 .
 N31 H31 O21 0.88 1.94 2.818(3) 173.1 2_655
 N41 H41 O51 0.88 2.00 2.831(3) 157.5 .
 N51 H51 O31 0.88 2.16 2.845(3) 134.4 3_665

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_diffn_measured_fraction_theta_full    0.944
_refine_diff_density_max               0.326
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data_7
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_atom_type_scatter_dispersion_imag
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'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'H' 'H' 0.0000 0.0000
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'Fe' 'Fe' 0.3463 0.8444
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'N' 'N' 0.0061 0.0033
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'O' 'O' 0.0106 0.0060
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'

# Crystal Data

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_symmetry_space_group_name_Hall        'C 2y'

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_symmetry_equiv_pos_as_xyz
'x, y, z'
'-x, y, -z'
'x+1/2, y+1/2, z'
'-x+1/2, y+1/2, -z'

_cell_length_a                          15.4790(3)
_cell_length_b                          9.3420(3)
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_cell_angle_beta       117.071 (2)
_cell_angle_gamma      90.00
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_cell_formula_units_Z  2
_cell_measurement_temperature 173 (2)
_cell_measurement_reflns_used 3809
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_exptl_crystal_size_min 0.15
_exptl_crystal_density_meas ?
_exptl_crystal_density_diffrn 1.305
_exptl_crystal_density_method 'not measured'
_exptl_crystal_F_000 592
_exptl_absorpt_coefficient_mu 0.575
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_exptl_absorpt_correction_T_min 0.8696
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_exptl_absorpt_process_details 'HKL Scalepack (Otwinowski & Minor
1997) '

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# EXPERIMENTAL DETAIL

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tube'
_diffrn_radiation_monochromator 'horizontally mounted graphite
crystal'
_diffrn_measurement_device '95mm CCD camera on \k-goniostat'
_diffrn_measurement_device_type 'KappaCCD (Bruker AXS-Nonius,
FR540C) '
_diffrn_measurement_method 'CCD rotation images, thick slices'
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_diffrn_standards_number ?
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_diffrn_standards_decay_% 'no decay'
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_refl_number_gt               3656
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_computing_cell_refinement    'HKL Scalepack (Otwinowski & Minor
1997)'
_computing_data_reduction
; HKL Denzo and Scalepack (Otwinowski & Minor 1997)
;
_computing_structure_solution
; 'SIR-97 (A. Altomare, G. Cascarano et al.)'
;
_computing_structure_refinement  'SHELXL-97 (Sheldrick, 1997)'
_computing_molecular_graphics
; Cameron (WinGX v1.70.01, 2005), Ortep (SHELXTL-NT v6.14, Bruker AXS)
;
_computing_publication_material  'WinGX v1.70.01 (L. J. Farrugia)'
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REFINEMENT DATA

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Refinement of F2 against ALL reflections. The weighted R-factor wR
and
goodness of fit S are based on F2, conventional R-factors R are
based
on F, with F set to zero for negative F2. The threshold expression
of
F2 > 2sigma(F2) is used only for calculating R-factors(gt) etc.
and is
not relevant to the choice of reflections for refinement. R-factors
based
on F2 are statistically about twice as large as those based on F,
and R-
factors based on ALL data will be even larger.
;

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_refine_ls_matrix_type            full
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_refine_ls_weighting_details
'calc w=1/[s2(Fo2)+(0.0279P)2+0.6901P] where
P=(Fo2+2Fc2)/3'
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_atom_sites_solution_hydrogens    geom
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_refine_ls_abs_structure_details  'Flack H D (1983), 1843 Friedel
pairs'
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_refine_ls_abs_structure_Flack      -0.002(12)
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_refine_ls_wR_factor_ref            0.0753
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_refine_ls_goodness_of_fit_ref      1.069
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ATOMIC COORDINATES AND DISPLACEMENT PARAMETERS

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O11 O 0.18569(9) 0.55835(14) -0.02474(14) 0.0288(3) Uani 1 1 d . . .
O12 O 0.13540(10) 0.74324(14) 0.17325(14) 0.0313(3) Uani 1 1 d . . .
O13 O 0.25441(10) 0.91279(14) 0.24656(13) 0.0280(3) Uani 1 1 d . . .
N11 N 0.02054(10) 0.56661(14) -0.14226(14) 0.0200(3) Uani 1 1 d . . .
H11 H -0.0284 0.6265 -0.1811 0.024 Uiso 1 1 calc . . .
N12 N 0.18035(11) 0.85584(16) 0.02912(15) 0.0232(3) Uani 1 1 d . . .
H12 H 0.2221 0.9153 0.0225 0.028 Uiso 1 1 calc . . .
C10 C -0.00352(15) 0.41951(19) -0.15196(19) 0.0200(4) Uani 1 1 d . . .
C11 C 0.05515(12) 0.2969(3) -0.13749(16) 0.0261(3) Uani 1 1 d . . .
H11A H 0.1218 0.2970 -0.1170 0.031 Uiso 1 1 calc . . .
C12 C -0.00472(18) 0.1745(2) -0.1595(2) 0.0305(5) Uani 1 1 d . . .
H12A H 0.0152 0.0778 -0.1564 0.037 Uiso 1 1 calc . . .
C13 C -0.09902(17) 0.2208(2) -0.1868(2) 0.0295(5) Uani 1 1 d . . .
H13 H -0.1531 0.1606 -0.2051 0.035 Uiso 1 1 calc . . .
C14 C -0.09864(15) 0.3734(2) -0.1819(2) 0.0234(4) Uani 1 1 d . . .
H14 H -0.1521 0.4331 -0.1960 0.028 Uiso 1 1 calc . . .
C15 C 0.10941(12) 0.62563(18) -0.08062(16) 0.0195(3) Uani 1 1 d . . .
C16 C 0.10767(10) 0.7894(2) -0.09278(14) 0.0208(3) Uani 1 1 d . . .
H16 H 0.0421 0.8254 -0.1105 0.025 Uiso 1 1 calc R . .
C17 C 0.12695(16) 0.8297(2) -0.2125(2) 0.0356(5) Uani 1 1 d . . .
H17A H 0.1271 0.9341 -0.2205 0.043 Uiso 1 1 calc R . .
H17B H 0.0760 0.7891 -0.2965 0.043 Uiso 1 1 calc R . .
H17C H 0.1901 0.7916 -0.1971 0.043 Uiso 1 1 calc R . .
C18 C 0.18621(12) 0.82982(17) 0.15184(18) 0.0218(3) Uani 1 1 d . . .
C19 C 0.29240(16) 0.8806(2) 0.3927(2) 0.0336(4) Uani 1 1 d . . .
C110 C 0.34269(19) 0.7374(3) 0.4214(3) 0.0506(6) Uani 1 1 d . . .
H11B H 0.2947 0.6618 0.3768 0.061 Uiso 1 1 calc R . .
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H11C H 0.3755 0.7206 0.5198 0.061 Uiso 1 1 calc R . .  
H11D H 0.3905 0.7368 0.3863 0.061 Uiso 1 1 calc R . .  
C111 C 0.2133(2) 0.8875(3) 0.4364(2) 0.0476(6) Uani 1 1 d . . .  
H11E H 0.1749 0.9746 0.3999 0.057 Uiso 1 1 calc R . .  
H11F H 0.2424 0.8889 0.5360 0.057 Uiso 1 1 calc R . .  
H11G H 0.1712 0.8035 0.4018 0.057 Uiso 1 1 calc R . .  
C112 C 0.3637(2) 1.0022(3) 0.4560(2) 0.0619(8) Uani 1 1 d . . .  
H11H H 0.4121 1.0004 0.4219 0.074 Uiso 1 1 calc R . .  
H11I H 0.3962 0.9913 0.5552 0.074 Uiso 1 1 calc R . .  
H11J H 0.3289 1.0937 0.4320 0.074 Uiso 1 1 calc R . .
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  _atom_site_aniso_U_12  
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O11 0.0201(6) 0.0228(6) 0.0406(7) -0.0029(5) 0.0111(6) 0.0056(5)  
O12 0.0310(7) 0.0319(7) 0.0351(7) -0.0037(5) 0.0186(6) -0.0128(5)  
O13 0.0308(7) 0.0237(6) 0.0249(6) -0.0017(5) 0.0085(6) -0.0101(5)  
N11 0.0187(7) 0.0153(6) 0.0240(7) -0.0009(5) 0.0080(6) 0.0007(5)  
N12 0.0230(7) 0.0202(7) 0.0280(7) -0.0031(6) 0.0131(6) -0.0067(6)  
C10 0.0246(9) 0.0131(8) 0.0214(8) -0.0023(7) 0.0098(7) -0.0002(7)  
C11 0.0320(8) 0.0182(7) 0.0307(7) -0.0051(9) 0.0167(7) -0.0008(9)  
C12 0.0418(13) 0.0172(10) 0.0328(11) -0.0080(8) 0.0174(10) -0.0019(9)  
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C15 0.0212(8) 0.0185(7) 0.0213(7) -0.0004(6) 0.0117(7) 0.0025(6)  
C16 0.0207(7) 0.0159(6) 0.0255(7) -0.0014(9) 0.0103(6) -0.0022(9)  
C17 0.0484(12) 0.0310(11) 0.0303(9) 0.0012(7) 0.0204(9) -0.0073(8)  
C18 0.0188(8) 0.0182(8) 0.0284(8) -0.0036(6) 0.0108(7) -0.0038(6)  
C19 0.0369(11) 0.0330(10) 0.0232(9) 0.0023(7) 0.0068(8) -0.0045(8)  
C110 0.0494(14) 0.0495(13) 0.0447(13) 0.0149(11) 0.0144(12) 0.0119(12)  
C111 0.0561(15) 0.0582(15) 0.0322(11) -0.0033(10) 0.0232(11) -  
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All esds (except the esd in the dihedral angle between two l.s.  
planes)  
are estimated using the full covariance matrix. The cell esds are  
taken  
into account individually in the estimation of esds in distances,  
angles  
and torsion angles; correlations between esds in cell parameters are  
only  
used when they are defined by crystal symmetry. An approximate  
(isotropic)  
treatment of cell esds is used for estimating esds involving l.s.  
planes.  
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```

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MOLECULAR GEOMETRY

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Fe1 C14 2.038(2) . ?
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Fe1 C10 2.0384(18) . ?
Fe1 C13 2.041(2) 2 ?
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Fe1 C12 2.048(2) 2 ?
Fe1 C12 2.048(2) . ?
Fe1 C11 2.0528(15) 2 ?
Fe1 C11 2.0528(15) . ?
O11 C15 1.227(2) . ?
O12 C18 1.224(2) . ?
O13 C18 1.343(2) . ?
O13 C19 1.475(2) . ?
N11 C15 1.344(2) . ?
N11 C10 1.415(2) . ?
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C17 H17C 0.9800 . ?
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C19 C112 1.515(3) . ?
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C110 H11C 0.9800 . ?
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C111 H11F 0.9800 . ?
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C112 H11H 0.9800 . ?
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C10 Fe1 C11 136.97(8) 2 . ?
C10 Fe1 C11 40.76(8) . . ?
C13 Fe1 C11 112.36(8) 2 . ?
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C15 N11 C10 127.59(15) . . ?
C15 N11 H11 116.2 . . ?
C10 N11 H11 116.2 . . ?
C18 N12 C16 122.06(14) . . ?

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C18 N12 H12 119.0 . . ?
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N11 C10 C14 121.42(17) . . ?
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C14 C10 C11 108.79(17) . . ?
N11 C10 Fe1 126.73(13) . . ?
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C16 C17 H17B 109.5 . . ?
H17A C17 H17B 109.5 . . ?
C16 C17 H17C 109.5 . . ?
H17A C17 H17C 109.5 . . ?
H17B C17 H17C 109.5 . . ?
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O12 C18 O13 125.27(17) . . ?
N12 C18 O13 110.52(14) . . ?
O13 C19 C110 108.76(19) . . ?
O13 C19 C111 111.33(18) . . ?
C110 C19 C111 112.8(2) . . ?
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C111 C19 C112 110.3(2) . . ?
C19 C110 H11B 109.5 . . ?

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C19 C110 H11C 109.5 . . ?
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Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.
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MOLECULAR GEOMETRY

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C12 C13 C14 108.5(3) . . ?
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N12 C16 C15 111.5(2) . . ?
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